

A photon is in a state:

$$\frac{1}{\sqrt{2}} |x_1\rangle + \frac{1}{\sqrt{2}} |x_2\rangle,$$

a quantum superposition of two position eigenstates. This state describes:

- A) one photon at x_1 and another at x_2 .
- B) a single photon at a position somewhere in between x_1 and x_2 .
- C) a single photon at a specific location, but we don't know the location since we haven't measured it yet.
- D) a single photon that does not have a definite location.

$$\begin{matrix} \bullet & & \bullet \\ X_1 & & X_2 \end{matrix}$$

An electron is in a state

$$\frac{1}{2}|X_1\rangle - \frac{\sqrt{3}}{2}|X_2\rangle$$

If we measure the electron's position, the result that we are most likely to find is

- A) X_1
- B) X_2
- C) $\frac{1}{2} X_1 - \frac{\sqrt{3}}{2} X_2$
- D) $\frac{1}{4} X_1 + \frac{3}{4} X_2$
- E) None of the above

$$\begin{array}{c} \bullet \\ X_2 \\ \bullet \\ X_1 \end{array}$$

An electron is in a state

$$\frac{1}{2}|X_1\rangle - \frac{\sqrt{3}}{2}|X_2\rangle$$

If we measure the electron's position, the result that we are most likely to find is

A) $X_1 \rightarrow \text{Prob. } \frac{1}{4} = \left|\frac{1}{2}\right|^2$

B) $X_2 \rightarrow \text{Prob. } \frac{3}{4} = \left|-\frac{\sqrt{3}}{2}\right|^2$

C) $\frac{1}{2}X_1 - \frac{\sqrt{3}}{2}X_2$

D) $\frac{1}{4}X_1 + \frac{3}{4}X_2 \rightarrow$ this is the average value if we did the experiment a large #

E) None of the above $\frac{3}{4}$ times.

Immediately repeated measurements of an electron's position give the same result. This implies that

- A) The electron's wavefunction generally stays the same when we do a measurement.
- B) The electron's wavefunction generally changes when we do a measurement.

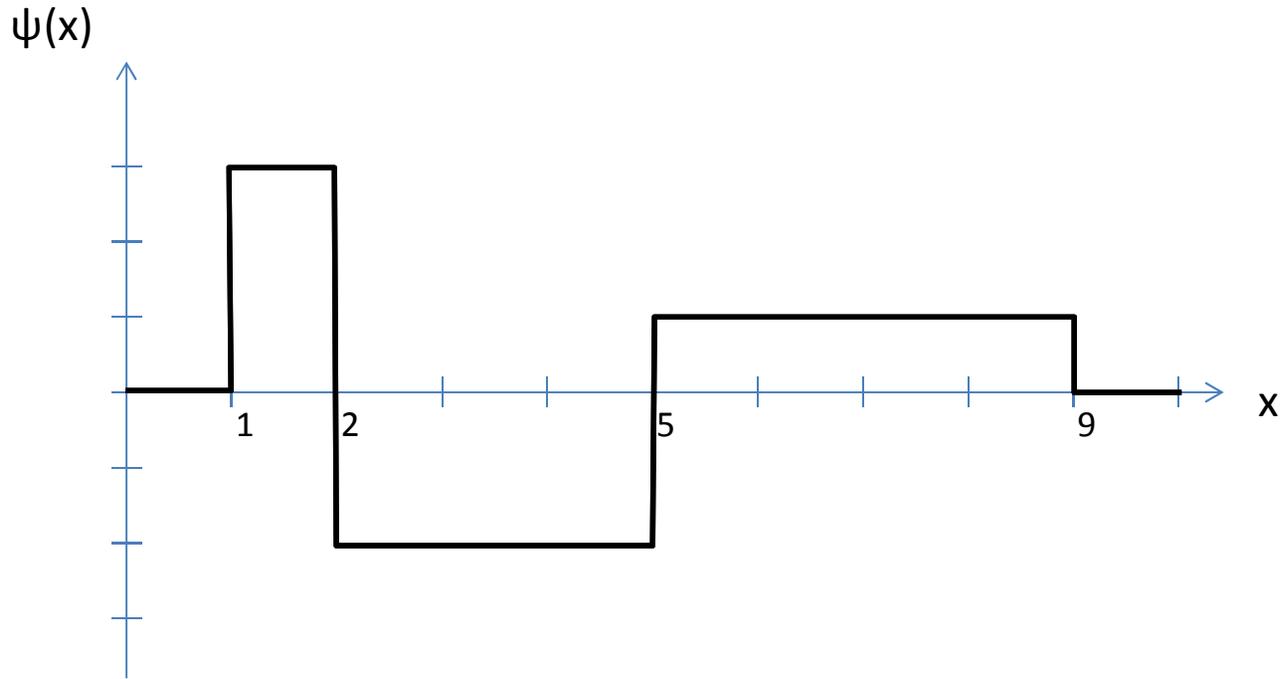
Immediately repeated measurements of an electron's position give the same result. This implies that

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Original wavefunction → superposition of position eigenstates,
→ position not predetermined

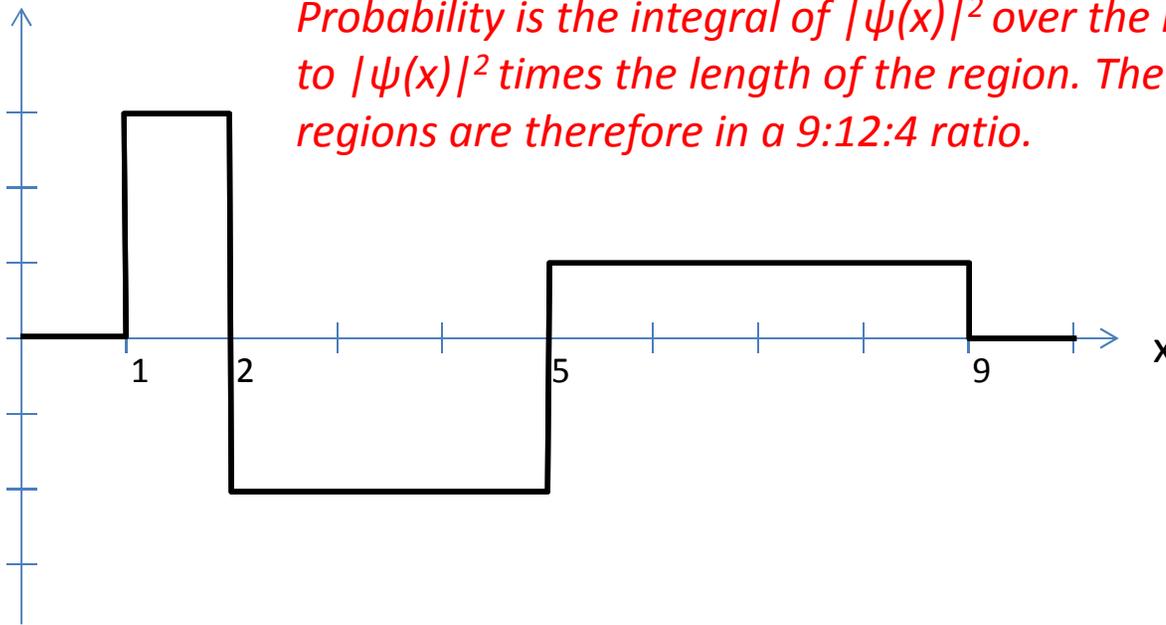
After measurement: know position
→ state = position eigenstate
∴ wavefunction has changed



The wavefunction for an electron in a one-dimensional wire is shown. If we measure the position, the electron is most likely to be found:

- A) Between 1 and 2
- B) Between 2 and 5
- C) Between 5 and 9
- D) All are equally likely
- E) The answer cannot be determined from the information given

$\psi(x)$



Probability is the integral of $|\psi(x)|^2$ over the region, so is proportional to $|\psi(x)|^2$ times the length of the region. The probabilities for the three regions are therefore in a 9:12:4 ratio.

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- B) Between 2 and 5**
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