The picture shows a wavefunction of an electron in a wire. If we measure the electron's position, we are most likely to find it:



- D) all three regions are equally likely
- E) how can the probability be negative?

The picture shows a wavefunction of an electron in a wire. If we measure the electron's position, we are most likely to find it:



between x2 and x3

D) all three regions are equally likely

E) how can the probability be negative?





For an electron with the wavefunction shown on the left, which is a likely wavefunction shortly after a measurement of position is made?

> the wavefunction will became sharply peaked around the position where the electron is found. B is wrong because the wavefunction is zero at this point.

The (not normalized) wavefunction of a particle is given by $\Psi(x) = e^{i(2\pi/h)px}$

Which of the pictures below best represents the probability density for finding the particle at point x?



The (not normalized) wavefunction of a particle is given by $\Psi(x) = e^{i(2\pi/h)px}$

Which of the pictures below best represents the probability density for finding the particle at point x?







If we measure the momentum of a particle, its wavefunction will typically

- A) stay the same
- B) became more localized
- C) became more spread out

If we measure the momentum of a particle, its wavefunction will typically

A) stay the same

B) became more localized

C) became more spread out

after you measure the momentum, the wavefunction must be in an eigenstate of momentum - a one-component pure wave. This is very spread out.