An electron is in a state (1/2)  $|3cm> - (i\sqrt{3}/2)|4cm>$ . (The lengths are distances along the x-axis from some agreed-upon origin). If we measure the electron's position, we are most likely to find a result of

A) 3cm

- B) 4cm
- C) (3/2 2i√3) cm
- D)  $(3/2 + 2\sqrt{3})$  cm
- E) anywhere between 3cm and 4cm

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A) 3cm

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prob of detection at |3cm> is  $|1/2|^2 = 1/4$ prob of detection at |4cm> is  $|i\sqrt{3}/2|^2 = 3/4$ the latter is bigger, so that's where the electron is most likely to be found

C) (3/2 - 2i√3) cm

D)  $(3/2 + 2\sqrt{3})$  cm

E) anywhere between 3cm and 4cm

An electron is in a state given by the wavefunction on the right. (darker region means  $\Psi$  is larger)

A measurement finds the electron at the spot marked with  $\Rightarrow$ 

Shortly after, the measurement of position is repeated. Where is the electron is most likely to be found second time around?



E) - impossible to predict, since everything is random

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A measurement finds the electron at the spot marked with  $\Rightarrow$ 

Shortly after, the measurement of position is repeated. Where is the electron is most likely to be found second time around? once the particle is found at the star it the first measurement, it must be detected there in all subsequent measurements



E) - impossible to predict, since everything is random

Immediately repeated measurements of an electron's position give the same result. This implies that

A) The electron's wavefunction generally stays the same when we do a measurement.

B) The electron's wavefunction generally changes when we do a measurement.

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When a measurement is made, the quantum state changes to an eigenstate of the measurement